
DESIGN OF ANAEROBIC BIOREACTOR FOR THE TREATMENT OF FIVE TONS PER DAY OF PETROLEUM SLUDGE

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ABSTRACT

Untreated petroleum sludge constitutes health hazard to personnel on exposure and environmental hazard on disposal. The treatment of sludge destroys disease pathogens present in the sludge. Under microbial substrate limiting Mesophilic anaerobic digestion, a bioreactor for the treatment of 5 tons per day of sludge was designed. Dimensions of the bioreactor were obtained by substituting kinetic and experimental data into design equations. The potential of anaerobic digestion for the treatment of sludge was proved by a reduction in Biochemical Oxygen Demand (BOD) of the sludge from 6,080 mg/L to 20.40 mg/L and Total Hydrocarbon Content (THC) from 57,000 ppm to 1500 ppm. Gas Chromatography and Mass Spectrophotometry (GC-MS) result showed a decrease in concentration of priority Polycyclic Aromatic Hydrocarbons (PAHs) from 37.1 mg/L to 0.32 mg/L Napthalene; 33.43 mg/L to 8.24 mg/L Anthracene and 33.97 mg/L to 9.86 mg/L Phenanthrene. Anaerobic digestion of 200 grams of sludge gave 10,500 m³ /d biogas, meaning that 1 gram of sludge could yield 840 m³ biogas for solids retention time of 16 days. The 183.50 m³ Volume; 15.93 m height; 3.83 m diameter bioreactor design provided could help Nigerian industries and environmental agencies construct and install bioreactors for the treatment of their sludge at affordable cost. This will enhance Nigerian content in bioreactor manufacture and operation. With a bioreactor anaerobic digestion plant as a process unit in every Nigerian Process industry and Waste Water Treatment Plant, the problem of sludge treatment and disposal according to Federal Environmental Protection Agency (FEPA) standards and regulations will be solved.

Keywords: *Bioreactor; Sludge; Priority toxicants; Anaerobic Digestion; Biochemical Oxygen Demand; Total Hydrocarbon Content.*

1.0. INTRODUCTION

Petroleum industries generate large quantities of sludge which is a major source of environmental pollution (Islam, 2015). Petroleum sludge constitutes health hazard to personnel on exposure and environmental hazard on disposal, if not properly treated before disposal. Besides this, it is an economic waste to dispose sludge without treatment as the treatment yields important products such as biogas, pharmaceuticals, fertilizers, etc. (Appels *et al.*, 2008). In this study, a Fed-batch anaerobic bioreactor for the treatment of petroleum sludge is designed so as to make the sludge nontoxic to personnel on exposure and harmless to the environment on disposal. Islam (2015) stated that petroleum sludges are hazardous wastes according to Environmental Protection Act and Hazardous Waste Handling Rules. These sludges cannot be disposed off as landfills even if they are de-oiled unless they are totally remediated. These sludges have to be treated and made harmless before disposal. Nigerian industries find it difficult to adhere to Federal Environmental Protection Agency (FEPA) standards and regulations for the treatment and disposal of sludge because they do not have bioreactors as an essential process unit in their process industries and waste water treatment plants. This is evidenced by the author's numerous visits to many process industries where sludges are heaped with the hope of incinerating them in a burn pit. In Nigeria, hazardous wastes (sludge) have been dumped in various places e.g. Abattoir in Rivers State. The Nigerian government is recently embarking on clean up of areas polluted with sludge e.g. Ogoni land in Rivers State. In Nigeria, the treatment and disposal of sludge imposes a major challenge to the oil and gas industry. This study has provided a solution to this problem as the relevant industries and environmental agencies can now make use of the design data to fabricate and install suitable bioreactors for the treatment of their sludge at affordable cost. The treatment of sludge in bioreactors will eliminate the odour, health and environmental hazard associated with disposal of untreated sludge. Moreover, biogas, a substitute for natural gas will be produced. This will enhance the realization of the Nigerian quest for self sufficiency in renewable energy. This is not a mere speculation as biogas sources are not yet fully utilized in Nigeria as it is in the advanced countries (Emberga *et al.*, 2014; Osai, 2012; Fasina & Simonyan, 2013; Temilade, 2008).

Kavitha & Pharm (2006) stated that continuous stirred tank bioreactors are constructed according to recognized standards as published by International Standards Organization and British Standards Institute. These dimensions take into account both mixing, effectiveness and structural consideration. A mechanically stirred tank bioreactor is filled with a rushton turbine type impeller.



Fig. 1: Pictorial view of a Bioreactor
Source: Wikipedia (2014)

Rao (2010) gave a generalized model for a Fed-batch bioreactor as in equation (1)

$$\frac{dC_i}{dt} = \frac{V(t)}{V_{Rc}} (C_{i,o} - C_i) + r_{fi} \quad (1)$$

Green & Perry (1997) stated that the sludge is first hydrolysed to become water soluble and then degraded to produce volatile organic acids primarily acetic acid and hydrogen. *Methanogenic* bacteria then split the acetic acid to methane and carbondioxide (biogas).

Impurities in the biogas are removed by pressure swing adsorption (PSA) on activated carbon. Since adsorption takes place at high temperature and pressure, desorption is achieved by depressurizing. Moisture is removed from the biogas by drying. The active site of the adsorbent retains water vapour and other pollutants thus decreasing adsorbent life, hence desorption is frequently carried out by depressurizing. Moreover, siloxanes are difficult to desorb from the adsorbent beds, so the adsorbent beds should be replaced regularly e.g. weekly. The biogas is dried, compressed and sent to storage.

Technical & Regulatory Guideline (2006) stated that *methanogenic* bacteria prefer a relatively neutral pH of 6.6 to 7.4 and not acidic conditions. If acid formation is excessive the activity of the *Methanogenic* bacteria can be inhibited.

It is good practice to destroy the volatile acids as quickly as they are produced otherwise the volatile acids build up and depress the pH and eventually inhibit the *methanogenic* bacteria. To prevent this occurrence feed to the digester should be as uniform as possible and at short intervals as possible.

Appels *et al.* (2008) gave the steps to sludge biodegradability as follows: Hydrolysis, Acidogenesis, Acetogenesis and Methanogenesis. Among this, Hydrolysis is the rate limiting step. This rate limiting hydrolysis is enhanced by biological, chemical and mechanical interventions to the sludge. These interventions result in lysis or disintegration of sludge cells hence transforming the sludge into a biodegradable material. Hence, a test for the rate of biodegradability of sludge species, results in experimental data which fit into equations developed from kinetic models.

Anaerobic digestion results in the conversion of the biodegradable sludge to methane, carbon dioxide and microbial cells. Volatile Suspended Solids (VSS) produced are quite low. Biochemical Oxygen Demand (BOD) is destroyed. Biogas produced range from 50 to 80 percent methane and 20 to 50 percent carbon (iv) oxide (CO₂) depending on the chemical characteristics of the sludge digested. The biogas produced is important for heat and power generation.

The pH should be maintained at 6.5 - 7.5 with the help of hydrochloric acid to avoid free ammonia toxicity. The use of nitric or sulphuric acid could result in significant operational problems.

2.0. MATERIALS AND METHODS

2.1. Materials

The following materials were used for the study:

Dissolved oxygen reagent bottles, Mineral water, 25mL density bottle, Hanna pH meter H196107, Muffle furnace LMF4, Thermospectronic spectrophotometer 4001/4, Oven, Gallenkamp incubator, Autoclave, Labtech anaerobic jar, Olympus microscope, Agilent gas chromatography and mass spectrophotometer (GC-MS) 7890 B, Spectr AA 55 B atomic absorption spectrophotometer, Petroleum sludge collected from a typical Petroleum Industry in Port Harcourt, Nigeria. *Methanogenic* Bacteria (*Methanobrevibacter*) isolated from the intestine of cow. Oxoid Anaero Gen TM AN 0035A gas Park used in labtech anaerobic Jar to create anaerobic condition, Winkler reagent A&B.

2.2. Methods

200 grams of the sludge was measured using a chemical balance and put into a beaker. 2 grams of *methanogenic methanobrevibacter* bacteria was pip potted and put into the sludge in the beaker after which the beaker was put into labtech anaerobic jar with improvise for gas collection point. Anaerobic condition was maintained with the help of oxoid Anaerobic Gen TM AN 0035A gas park and catalyst. The anaerobic jar was corked airtight and kept in a Gallenkamp incubator maintained at 37 °C (mesophilic) for a solids retention time of sixteen

days. After performing a ten fold serial dilution; the Total Anaerobic Bacterial Count in CFU/g was calculated using the equation (2).

$$TABC = \frac{1}{DF} \times \text{Average of plate bacterial count} \times \frac{1}{\text{Volume Correction factor}} \quad (2)$$

The percent Volatile Suspended Solids (VSS) measured using gravimetric method was calculated using equation (3).

$$\text{Volatile Suspended Solids} = \frac{\text{Weight of Volatile residue}}{\text{Weight of residue}} \times \frac{100}{1} \quad (3)$$

The Total Hydrocarbon Content (THC) measured using spectrophotometric method was calculated using equation (4).

$$THC = \frac{\text{Absorbance of sludge} \times \text{Gradient of standard graph}}{\text{weight of sample diluted in 100mL} \times \frac{1}{\text{Dilution Factor}}} \quad (4)$$

The Biochemical Oxygen Demand (BOD) measured using the Winkler's method was calculated using the equation (5).

$$BOD_5 = \frac{DO_{initial} - DO_{Final}}{\text{DilutionFactor}} \quad (5)$$

The concentration of Polycyclic Aromatic Hydrocarbons (PAHs) in the sludge and in the biosolids produced from anaerobic digestion of the sludge was measured using Gas Chromatography – Mass Spectrophometry (GC-MS).

2.2.1. Kinetic Models

The following kinetic models were derived in Sampson (2016):

For the Biomass

$$\frac{dX_1}{dt} = D (X_{1,0} - X_1) + \frac{\mu_m X_1 X_2}{k_m + X_2} \quad (6)$$

For the Sludge

$$\frac{dX_2}{dt} = D (X_{2,0} - X_2) - \frac{\mu_m X_1 X_2}{Y(k_m + X_2)} \quad (7)$$

Comparing it with equation (1)

$$\frac{dC_i}{dt} = \frac{V_t}{V_{RC}} (C_{i,0} - C_i) + r_{fi}$$

$$\text{Dilution rate} = \frac{V_t}{V_{RC}} = D$$

For Biomass:

$$r_{fi} = + \frac{\mu_m X_1 X_2}{K_m + X_2} - k_d X_1 \quad (8)$$

The positive sign depicts that the biomass is multiplying.

For Sludge:

$$r_{fi} = -\frac{\mu_m X_1 X_2}{\gamma(K_m + X_2)} \quad (9)$$

This gives the rate of reaction. The negative sign depicts that the sludge is biodegrading. Note that $K_d X_1$ is not included in the r_{fi} for sludge because K_d , the endogenous respiration coefficient is zero for the sludge.

$\frac{dx_2}{dt} = \frac{\mu_m X_1 X_2}{Y(k_m + x_2)}$ give the rate of reaction r_{fi} . Therefore equation (1) can be written in another form as:

$$\frac{dX_2}{dt} = D(X_{2,0} - X_2) - \frac{\mu_m X_1 X_2}{\gamma(K_m + X_2)}$$

2.2.2. Design Equations

In Sampson (2018), Equation (10) was derived for volume of the bioreactor.

$$V_R = \frac{NA_o}{tK_m X_{2,0}} \int_0^{X_A} \frac{dX_A}{(1-X_A)(1+E_A X_A)} \quad (10)$$

Where V_R is the volume of the Bioreactor

Yield coefficient was calculated using equation given in Econotres (2014)

$$\text{Yield Coefficient (Y)} = \frac{mgVSS}{mg BOD} \quad (11)$$

Appels *et al.* (2008) gave equation for Net mass of cell tissue produced per day:

$$P_x = \frac{Y_{ES_0}}{1+k_d \theta_c} \quad (12)$$

Volume of Biogas produced was calculated using equation (13)

$$V_{CH_4} = (0.35)(S_0 - s)(Q) \left(\frac{10^3 \text{g}}{\text{kg}^{-1}} \right) - 1.42P_x \quad (13)$$

Tapobrata (2011) gave the following equations for bioreactor fundamental dimensions:

Height of Bioreactor

$$h = \frac{4V_R}{\pi D^2} \quad (14)$$

Impeller Diameter of Bioreactor

$$\frac{D_i}{D} = 0.3 \quad (15)$$

$$D_i = D \times 0.3$$

Width of Baffle

$$(0.08 - 0.1)D_i \quad (16)$$

Length of Jacket

$$J_L = \frac{1}{4} H \quad (17)$$

Vertical Distance between adjacent stirrer blades

$$\frac{I_L}{D_i} = 1.2 \quad (18)$$

$$\text{Poison ratio} = \frac{\text{hemispherical thickness}}{\text{Cylinder thickness}} \quad (19)$$

$$\text{Volume of hemisphere} = \frac{2}{3} \pi \times r^3 \div 2 \quad (20)$$

$$\text{Volume of conical bottom} = \frac{1}{8} V_R \quad (21)$$

$$\text{Height of conical bottom} = \frac{1/8 V_R \times 3}{\pi r^2} \quad (22)$$

3.0. RESULTS AND DISCUSSION

3.1. Data Obtained

Table 1: Experimental Data Obtained

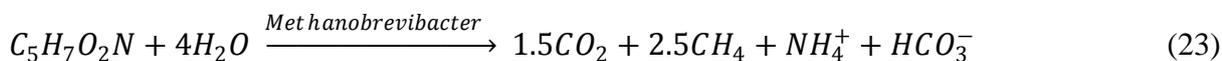
| Parameter | Value |
|---|---|
| Initial concentration of Biomass ($X_{1,0}$) | 1.0 mg/L |
| Initial concentration of sludge ($X_{2,0}$) | 0.3383 mg/L |
| Volatile Suspended Solids (VSS) | 99.6 % |
| Biochemical oxygen demand (influent sludge) | 6080 mg/L |
| Biochemical oxygen demand (effluent biosolids) | 20.40 mg/L |
| Yield coefficient (Y) | 0.016 |
| Maximum specific growth rate (μ_m) | 0.0738hr ⁻¹ |
| Dilution rate (D) or space velocity | 0.18hr ⁻¹ |
| Total hydrocarbon content (influent sludge) | 57000 ppm |
| Total hydrocarbon content (effluent biosolids) | 1500 ppm |
| Negative of log of Hydrogen ion Concentration (pH) | 6.5 |
| Net mass of cell tissue produced per day | 62.5 kg/d |
| Volume of biogas produced | 10,500 m ³ /d |
| Polycyclic aromatic hydrocarbons (influent sludge) | Thirties (mg/L) |
| Polycyclic aromatic hydrocarbons (effluent biosolids) | Units (mg/L) |
| Monods constant (K_m) | 0.02 kmol/m ³ |
| Endogenous respiration coefficient (K_d) | 0.025 d ⁻¹ |
| Total Anaerobic Bacterial Count (TABC) | Day 1: 0.129 × 10 ⁹ ; Day 2: 102.1 × 10 ⁷ ; Day 5: 2020 × 10 ⁹ ; Day 16: 7.199 × 10 ⁹ (CFU/g) |

The result of the TABC indicates that micro-organisms multiplied in the biological reaction. The microbial digestion was terminated after 16 days at the falling rate phase when most of the micro-organisms must have died. BOD and THC decrease with sludge biodegradation. BOD and THC can therefore be used as a measure of sludge biodegradation. The concentration of naphthalene, anthracene and phenanthrene in the untreated sludge reduced from 37.1 mg/L to 0.32 mg/L Naphthalene; 33.43 mg/L to 8.24 mg/L Anthracene; 33.97 mg/L to 9.86 mg/L Phenanthrene after the sludge treatment. According to Owabor & Owhiri (2011), Naphthalene, Phenanthrene and Anthracene are used as representative of the polycyclic aromatic hydrocarbons as they are in the priority toxicant list of EU and USEPA due to their Mutagenic and carcinogenic properties.

A reasonable volume of biogas can be produced in substantial amount from anaerobic digestion of the sludge as shown in analysis of equation (13). 200 g sludge yielded 10,500 m³/d biogas for 16 days. Considering 200 days operation per year, 1g sludge would yield 10,500/200=52.5 m³/d biogas. For 16 days, 52.5×16=840 m³ biogas from 1g of sludge.

3.2. Sizing the Bioreactor

Quist (2007) gave equation (23):



Where ε_A is given by

$$\begin{aligned} \Sigma CR &: 1 + 4 = 5 \\ \Sigma CP &: 1.5 + 2.5 + 1 + 1 = 6 \\ \varepsilon_A &= \frac{\Sigma CP - \Sigma CR}{\Sigma CR} = \frac{6-5}{5} \\ \varepsilon_A &= \frac{1}{5} = 0.2 \end{aligned}$$

Levenspiel (2001) gave formula for conversion:

$$X_A = 1 - \frac{C_A}{C_{A0}} \quad (24)$$

$$X_A = 1 - \frac{X_2}{X_{2,0}} \quad (25)$$

$$\begin{aligned} X_{2,0} &= 0.3383 \text{ mg/L} \\ X_2 &= 0.228893 \text{ mg/L} \end{aligned}$$

Note that treatment of sludge is a slow reaction, hence conversion must be low.

$$\begin{aligned} X_A &= 1 - \frac{0.2289}{0.3383} \\ X_A &= 1 - 0.6766 \\ X_A &= 0.3234 \end{aligned}$$

Coulson & Richardson (1991) gave the Monods constant

$$K_m : 0.02 \text{ mol/m}^3$$

t: solids retention time, 16 days

C_{A0} : initial concentration of sludge

$$C_{A0} = X_{2,0} = 0.3383$$

N_{A0} : Number of moles of sludge

$$\text{Molar mass of } C_5 H_7 O_2 N: (12 \times 5) + (1 \times 7) + (16 \times 2) + (14 \times 1) = 60 + 7 + 32 + 14 = 113$$

Plant Capacity: Given a Plant capacity of 5 Tons per day. Knowing that 1 Ton is equivalent to 1000 kg, So 5 Tons is equivalent to 5000 kg which is the mass of sludge digested per day.

$$\begin{aligned} &= \frac{\text{Mass}}{\text{MolarMass}} = \frac{5000\text{kg}}{113} \\ &= 44.24778761 \text{ kmol/day} \\ &= 44.24778761 \times 10^3 \text{ moles/day} \\ &= 44,247.78761 \text{ moles/day} \end{aligned}$$

Rewriting equation (10) replacing $X_{2,0}$ with C_{A0}

$$V_R = \frac{N_{A0}}{C_{A0} \cdot t \cdot K_m} \int_0^{X_A} \frac{dX_A}{(1-X_A)(1+\varepsilon_A X_A)} \quad (26)$$

Substituting $X_{2,0}$ for C_{A0} in equation (26) give:

$$V_R = \frac{N_{A0}}{X_{2,0} t K_m} \int_0^{0.3234} \frac{dX_A}{(1-X_A)(1+\varepsilon_A X_A)} \quad (27)$$

$$V_R = \frac{44,247.78761}{0.3383 \times 16 \times 0.02} \int_0^{0.3234} \frac{d(0.3234)}{(1-0.3234)(1+0.2 \times (0.3234))}$$

$$X_A = 0.3234$$

$$K_m = 0.02 \text{ mol/m}^3$$

$$C_{A0} = 0.3383 \text{ mg/L}$$

$$t = 16 \text{ days}$$

$$N_{A0} = 44,247.78761 \text{ moles/day}$$

$$\varepsilon_A = 0.2$$

Using MathCAD Software

$$V_R = 1.835 \times 10^5 \text{ Litres}$$

$$(1000 \text{ Litres} = 1 \text{ m}^3)$$

$$\text{Therefore: } V_R = \frac{1.835 \times 10^5}{1000} = \frac{1,83500.00}{1000} = 183.50 \text{ m}^3$$

$$\text{Volume of Bioreactor} = 183.50 \text{ m}^3$$

3.3. Optimization

Coulson & Richardson (2009) gave equation for purchased equipment cost.

$$PEC = a + b s^n \times MF \quad (28)$$

Where:

PEC is the purchased equipment cost. a & b , cost constants peculiar to the equipment.

s , the characteristics size parameter
 n , the index characteristic of equipment size.

Recall that:

$a = 28,000$ in 2010

That is, 28,000 multiplied by 1.96 in 2018 is equal to 54,880

Range for b : 0 to 104,000

s is same for 2018 as it was in 2010

$$s = 0.5$$

n is same for 2018 as it was in 2010.

$$n = 0.8$$

MF, materials factor for stainless steel = 1.3. Multiplying this by 1.96 in 2018 gives 2.548
 ≈ 2.55 in 2018

Table 2: Optimization of Bioreactor Diameter

PEC for the bioreactor in the year 2018 gave £ 148,906.4103.

| Diameter D (m) | a | b | s | bs | $a + b$ | $a + bs$ | n | bs^n | MF | $a + bs^n$ | Cost $a + bs^n$ MF |
|----------------------|--------|---------|-----|-----------|---------|-----------|-----|----------|------|------------|-----------------------|
| 0.5 | 6,875 | 12,985 | 0.5 | 6,492.5 | 19,860 | 13,367.50 | 0.8 | 1,121.84 | 2.55 | 7996.84 | 20,391.93308 |
| 1.0 | 13,750 | 25,970 | 0.5 | 12,985.0 | 39,720 | 26,735 | 0.8 | 1,953.23 | 2.55 | 15,703.23 | 40,043.23854 |
| 1.5 | 20,625 | 38,955 | 0.5 | 19,477.50 | 59,580 | 40,102.5 | 0.8 | 2,701.63 | 2.55 | 23,326.64 | 59,482.91762 |
| 2.0 | 27,500 | 51,940 | 0.5 | 25,970.0 | 79,440 | 53,470 | 0.8 | 3,400.77 | 2.55 | 30,900.77 | 78,796.96947 |
| 2.5 | 34,375 | 64,925 | 0.5 | 32,462.50 | 99,300 | 66,837.5 | 0.8 | 4,065.42 | 2.55 | 38,440.42 | 98,023.0746 |
| 3.0 | 41,250 | 77,910 | 0.5 | 38,955.0 | 119,160 | 80,205 | 0.8 | 4,703.82 | 2.55 | 45,953.82 | 117,182.2375 |
| 3.5 | 48,125 | 90,895 | 0.5 | 45,447.50 | 139,020 | 93,572.5 | 0.8 | 5,321.18 | 2.55 | 53,446.18 | 136,287.7605 |
| 4.0 | 55,000 | 103,880 | 0.5 | 51,940.0 | 158,880 | 106,940 | 0.8 | 5,921.09 | 2.55 | 60,921.09 | 155,348.7758 |

Interpolation for the Diameter of the Bioreactor at £ 148,906.4103 gave 3.83 meters.

Table 3: Bioreactor Fundamental Dimensions

| Parameter | Value |
|---|--------------|
| Volume of the Bioreactor (V_R) | 183.50 m^3 |
| Diameter of the Bioreactor (D) | 3.83 m |
| Height of Bioreactor (H) | 15.93 m |
| Impeller Diameter of Bioreactor (D_i) | 1.149 m |
| Width of Baffle (W_b) | 11.49 cm |
| Length of Jacket (J_L) | 3.9825 m |
| Vertical distance between adjacent stirrer blades (I_L) | 1.3799 m |
| Radius of Hemispherical head | 1.915 m |
| Height of Hemispherical head | 1.0 m |
| Radius of conical bottom | 1.915 m |
| Height of conical bottom | 6.0 m |

3.4. Design Features

3.4.1. Material of Construction

Stainless steel does not easily get corroded.

3.4.2. Wall Thickness

Coulson & Richardson (2009) gave 14 mm as the minimum thickness for a vessel of diameter 3.5 m – 4.0 m.

Wall thickness of bioreactor = 14 mm.

3.4.3. Vessel Pressure

Coulson & Richardson (2009) gave a pressure of 1 bar for vessels operated anaerobically. Pressure = 1 bar.

3.4.4. Spacing between the jacket and vessel wall

The spacing between the Jacket and vessel wall is 50 mm (Coulson & Richardson, 2009). Spacing must be high enough to enhance insulation.

3.4.5. Thickness of Jacket

Thickness of Jacket or lagging is 50 mm. Lagging must be thick enough to enhance heat conservation.

3.4.6. Lagging Material: Foam glass (Cellular glass made by fusing powdered glass with Carbon particles).

Foam glass is dense enough to allow for attachment of fittings and sealing and does not absorb moisture.

3.4.7. Cover for Lagging: Aluminum metal sheet (0.5 mm thick). Aluminum is cheap enough and is non corrosive.

3.4.8. Support: Concrete pillars.

Concrete pillars cannot conduct electricity.

3.4.9. Stirrer Driver: Rushton turbine.

Rushton turbine is more robust and has higher efficiency than electrical motors.

3.4.10. Hemispherical Head

3.4.10.1. Radius of hemispherical head

Radius of cylinder = 1.915 meters, and equal to the radius of hemispherical head.

3.4.10.2. Height of Hemispherical head

Take volume of hemisphere to be $\frac{1}{3}$ volume of cylinder.

Volume of hemisphere (half of a sphere) = $\frac{2}{3} \times \pi \times r^3 \div 2$

Radius of the hemisphere = 1.915 metres

Therefore:

Height of the hemisphere = $\frac{1.915}{2} = 0.9575 \approx 1.0 \text{ metres}$

A hemispherical head is the strongest shape, capable of resisting about twice the pressure of a torispherical head of the same thickness (Coulson & Richardson, 2009).

3.5. Conical Bottom

3.5.1. Radius of conical bottom

Radius of Conical Bottom is same as radius of cylinder = 1.915 *metres*.

Volume of the conical bottom is one eighth the volume of the cylinder.

Volume of cylinder = 183.50 m^3 .

3.6. Height of Conical Bottom

Conical bottom is used to facilitate the smooth flow and removal of solids from a process equipment (Coulson & Richardson, 2009).

Volume of the conical bottom is one eighth the volume of the cylinder.

Volume of cylinder = 183.50 m^3

Volume of conical bottom: $\frac{1}{8} \times 183.50 m^3 = 22.9375 m^3$

Volume of conical bottom = $\frac{1}{3} \pi r^2 h$ (29)

$$22.9375 = \frac{1}{3} \times 3.142(1.915)^2 h$$

$$h \approx 6.0 \text{ metres}$$

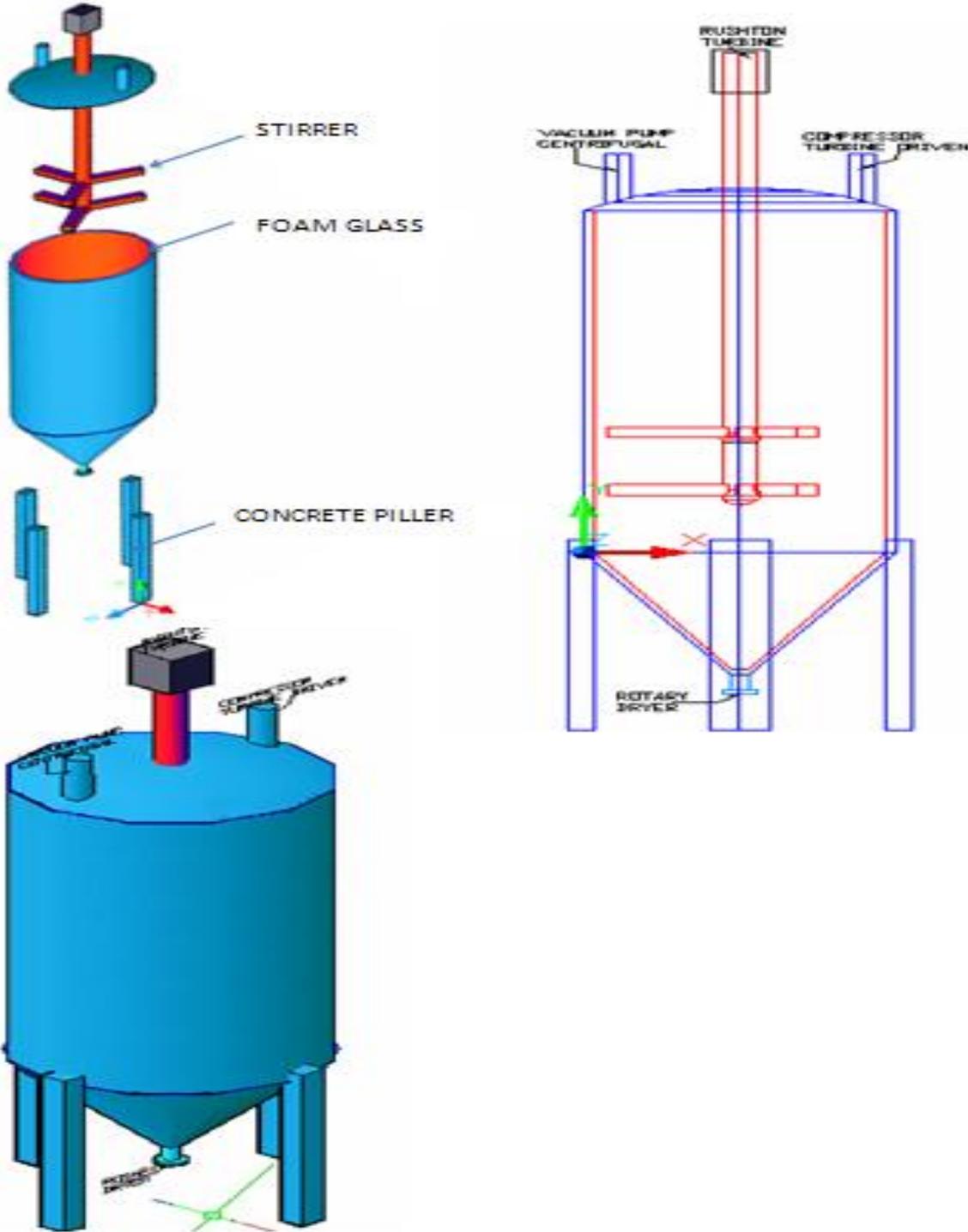


Fig. 2: Exploded views and 3-Dimensional view of the Bioreactor

A bioreactor of volume 183.50 m^3 , diameter 3.83 metres and height 15.93 metres can be constructed as shown in figure 2. However, the bioreactor can be scaled-up or scaled-down or remodified to meet the needs of an industry or agency.

4.0. CONCLUSION

A bioreactor with the following fundamental dimensions: Volume 183.50 m^3 ; diameter 3.83 metres; height 15.93 metres has been designed. Other dimensions of the designed bioreactor are: impeller diameter 1.149 metres; width of baffle 11.49 cm; length of jacket 3.9825 metres; vertical distance between adjacent stirrer blades 1.3788 metres; wall thickness 14 mm; spacing between jacket and vessel wall 50 mm; thickness of jacket 50 mm; Radius of hemispherical head 1.915 metres; Height of hemispherical head 1.0 metre; Radius of conical bottom 1.915 metres; Height of conical bottom 6.0 metres. With this design, the problem of sludge treatment in accordance with Nigerian FEPA standards and regulations for the treatment and disposal of sludge has been solved. The relevant industries and environmental agencies can now fabricate and install bioreactors for the treatment of their sludge at affordable cost. This could enhance Nigerian content in bioreactor manufacture and operation. The potential of anaerobic digestion for the treatment of petroleum sludge has been proved by a decrease in the concentration of polycyclic aromatic hydrocarbons. The priority toxicants in the sludge (Naphthalene, Phenanthrene and Anthracene) decreased in concentration from thirties to units. The biochemical oxygen demand decreased from 6080 mg/L to 20.40 mg/L and the total hydrocarbon content of the sludge from 57000 ppm to 1500 ppm. Hence anaerobic digestion helps to transform the toxic petroleum sludge to harmless biosolids. It was found that the treatment of 5 Tons per day of petroleum sludge could yield $10,500\text{ m}^3$ per day of biogas; hence, one gram of the sludge could yield 840 m^3 biogas for a solids retention time of sixteen days. Biogas, being a renewable energy source and a better substitute for natural gas, anaerobic digestion of petroleum sludge could enhance the realization of the Nigerian quest for self sufficiency in renewable energy. The plant should be located as a process unit in every Nigerian process industry, waste water treatment plants and sites where sludges are dumped.

5.0. RECOMMENDATIONS

It is recommended that:

- (i) Local design, construction and installation of the anaerobic bioreactor be carried out using the design data provided. Considering that Nigerian industries and environmental agencies could not treat their sludge because of high treatment cost and unavailability of design data, this could help Nigerian industries and environmental agencies treat their sludge at affordable cost.
- (ii) With the present call for amendment of the Nigerian Petroleum Industry law, every Nigerian process industry should have an anaerobic bioreactor unit as one of its process units. This will eliminate disposal of untreated toxic wastes (sludge) into the environment with its consequent health hazards on humans, pollution of soil, surface water, inland water bodies and death of aquatic life.

6.0. REFERENCES

- Appels, L., Jan Baeyens, Jan Degreve & Raf D. (2008). Principles and potential of the anaerobic digestion of waste activated sludge. *Progress in Energy and Combustion Science*, 30(6), 755-781. Available: www.elsevier.com. Accessed: 10th April, 2017.
- Coulson, J.M. & Richardson, J. F. (1991). *Chemical Engineering*. Volume 3, (3rd Edition), Oxford: Butterworth-Heinemann, an Imprint of Elsevier.
- Coulson, J.M., Richardson J.F. & Sinnott, R.K. (2009). *Chemical Engineering*. vol.6, 5th edition, New York: Pergamon Press.
- Econotres, D. (2014). Biological Kinetic parameter estimation. *Elsevier science*, 1(1), 14-18. Available: www.econotres.durep174.10007, Accessed: 11th April, 2017.
- Emberga, T., Obasi, I., Omenikolo A.I & Nwigwe C. (2014). Renewable Energy Potentials and Production in Nigeria. *Standard Research Journal*, 2(3), 55-59. Available: www.standresjournals.org. Accessed: 1st June, 2017.
- Green, D.W & Robert H. P. (1997). *Chemical Engineers' Handbook*. (Seventh Edition), Singapore: McGrawhill Book Company.
- Islam, B. (2015). Petroleum Sludge, its treatment and disposal: a review, *International Journal of Chemical Science*, 13(4), 1584-1602. Available: www.sadgunipublications.com, Accessed : 25th July, 2017.
- Kavitha, R. & Pharm M. (2006). *Design of fermenter*. USA: SRM University Press. Available: www.srmuniv.ac.in. Accessed: 21st June, 2017.
- Levenspiel, O. (2001). *Chemical Reaction Engineering*. Third Edition, Singapore: John Wiley and Sons Inc
- Ossai, O.S. (2012). *Evaluation of Gasoline Generator Modified for Biogas Utilization*. Nigeria: University of Nigeria, Nsukka.
- Owabor, C.N. & Owhiri, E. (2011). Utilization of perinwinkle shell carbon in the removal of polycyclic aromatic hydrocarbons from refinery waste water. *Journal of Nigerian society of chemical Engineers* '26(2), 121-130.
- Quist, U. (2007). *Handbook of Biological Waste Water Treatment, Design and Optimisation of activated sludge systems*. Netherlands: Utigerij Quist Publishing Company.
- Rao, D.G. (2010). *Introduction to Biochemical Engineering*. 2nd edition, New Delhi: Tata McGraw Hill Education Private Limited.
- Sampson, I.E. (2016). Modeling a Biochemical Reactor for the Treatment of Sludge. *International Society of Comparative Education, Science and Technology (ISCEST) Conference Journal*, 3(2), 180-209. Available: www.journal.iscest.org
- Sampson, I.E. (2018). *Design of a Bioreactor for the Treatment of Sludge*. 1st editon, European Union: Lambert Academic Publishing. an imprint of Omniscryptum. Available: www.morebooks.shop

Simonyan, K.J & Fasina, O. (2013). Biomass Resources and Energy Potentials in Nigeria. *African Journal of Agricultural Research academic Journals*, 8(1), 4975 – 4989. Available: www.academicjournals.org. Accessed: 1st August, 2017.

Tapobrata, P. (2011). *Bioreactors: Analysis and Design*. 1st edition, New Delhi: Tata McGraw Hill Education Private Limited.

Tchobanoglous, G., Franklin Burton L. & David Stensel H. (2004). *Wastewater Engineering, Treatment and Reuse*. 4th edition, New York: Mcgraw-hillcompany.

Technical & Regulatory Guideline. (2006). *Characterization, Design, Construction and Monitoring of Bioreactor Landfills*. USA: The Interstate Technology of Regulatory Council alternative landfill technologist team. Available: www.itrcweb.org. Accessed: 27th July, 2017.

Temilade, S. (2008). *Status of Renewable Energy Policy and Implementation in Nigeria*. United Kingdom: University of Nottingham

Wikipedia. (2014). *Bio-Chemical Engineering*. USA: The Free Encyclopedia. Available: www.google.com. Accessed: 1st May, 2017.

7.0. NOMENCLATURE

| Symbol | Definition | Unit |
|-----------|--|--------------------------------|
| C_i | Inlet concentration | kmol/m^3 |
| $C_{i,0}$ | Initial concentration of component | kmol/m^3 |
| D_i | Impeller diameter | m |
| E | Efficiency of sludge utilization | % |
| F_{AO} | Molar Feed rate | mols^{-1} |
| H | Height of the Bioreactor | m |
| I_L | Vertical distance between adjacent stirrer blades | m |
| J_L | Length of the jacket | |
| K_m | Monods constant | kmol/m^3 |
| k_d | Endogenous respiration coefficient or specific maintenance rate | (d^{-1}) |
| m | Mass of dry solids | kg |
| N_{AO} | Number of moles | mols/day |
| P_s | Percent solids expressed as decimal | |
| P_x | Net mass of cell tissue produced per day | kg/d |
| Q | Flow rate of methane | m^3/d |
| R | Gas constant, | kJ/kmol K |
| r_{fi} | Rate of reaction of component i | $\text{mgL}^{-1}\text{s}^{-1}$ |
| S | Biological carbonaceous oxygen demand (BCOD) in the effluent biosolids | mg/L |
| S_D | Side depth of Bioreactor | m |
| S_0 | Biological carbonaceous oxygen demand (BCOD) in the influent Sludge | mg/L |

| | | |
|------------|--|-----------|
| t | Time | days |
| V_R | Volume of Bioreactor | m^3 |
| V_{RC} | Culture Volume | m^3 |
| $V(t)$ | Volumetric Feedrate at time t | m^3/d |
| V_{CH_4} | Volume of methane produced | m^3/d |
| X_A | Conversion | |
| X_1 | Biomass concentration | mg/L |
| $X_{1,0}$ | Initial concentration of biomass | mg/L |
| X_2 | Sludge concentration | mg/L |
| $X_{2,0}$ | Initial concentration of Sludge | mg/L |
| Y | Yield coefficient given as mass of sludge or biomass produced per unit biosolids removed | |
| Z | Depth of bed | m |
| θ_c | Mean cell residence time | days |
| μ_m | Maximum specific growth rate or half minimal velocity concentration | hr^{-1} |
| ρ | Density | kg/m^3 |
| ρ_w | Specific weight of water | kg/m^3 |

Abbreviation

BOD
CFU
DF
DO
EPA
EU
FAO
FEPA
GC – MS

MF
PAHs
PEC
SRT
TABC
THC
USEPA
VSS
WWTP

Definition

Biological Oxygen Demand
Colony Forming Unit
Dilution Factor.
Dissolved Oxygen
Environmental Protection Agency
European Union
Food and Agricultural Organization
Federal Environmental Protection Agency
Gas Chromatography and Mass Spectrophotometry
Materials Factor
Polycyclic Aromatic Hydrocarbons
Purchased Equipment Cost
Solids Retention Time
Total Anaerobic Bacterial Count
Total Hydrocarbon Content
United States Environmental Protection Agency
Volatile Suspended Solids
Waste Water Treatment Plant